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Magnesiothermic reduction to produce porous silicon for lithium-ion batteries

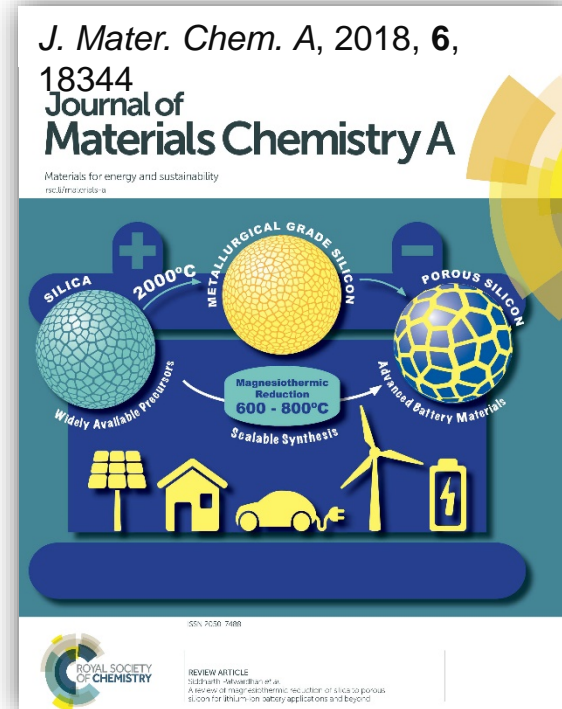
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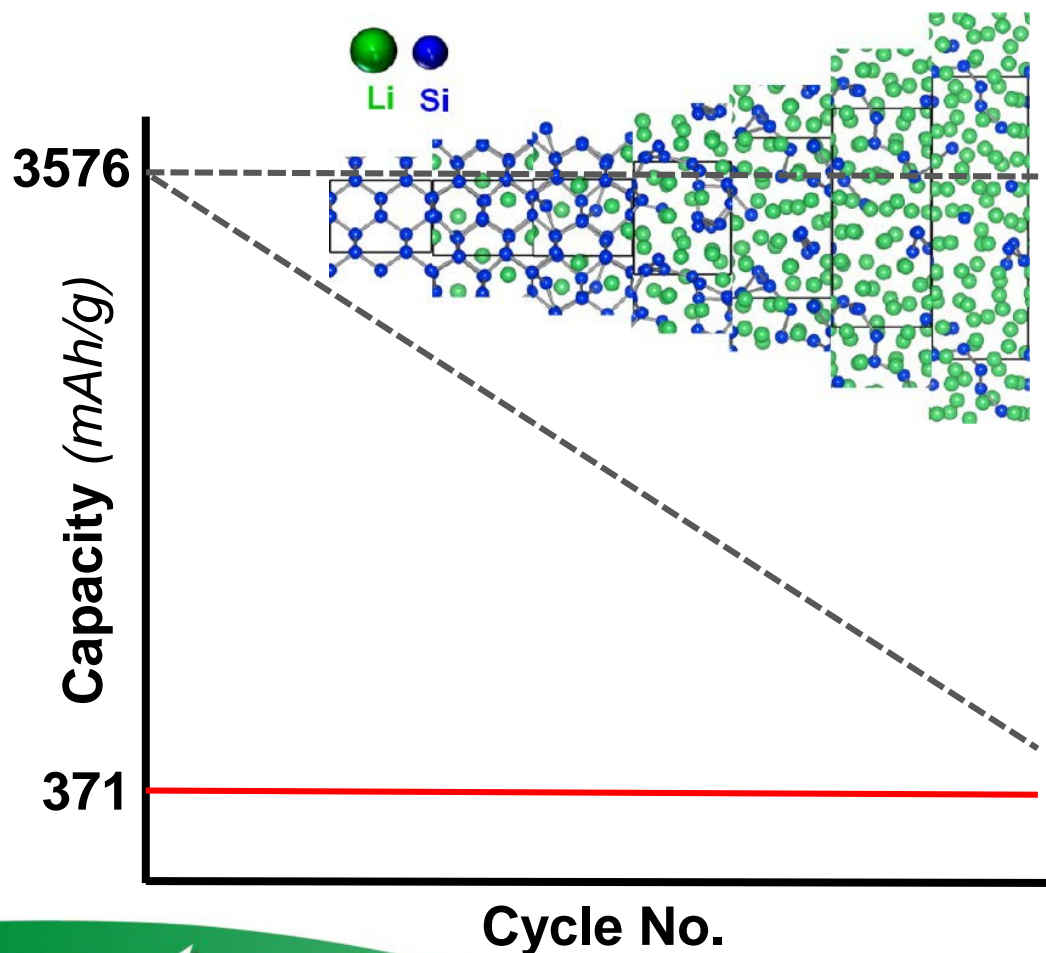
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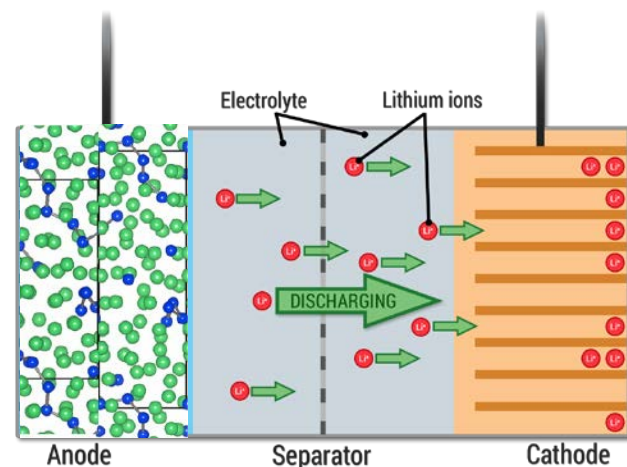
EFNL 249, ACS San Diego, 27/8/19

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Silicon vs conventional anode materials



Li_{3.75}Si (Li₁₅Si₄)
Gravimetric capacity **3576 mAh/g**
Volumetric capacity **2190 mAh/cm³**



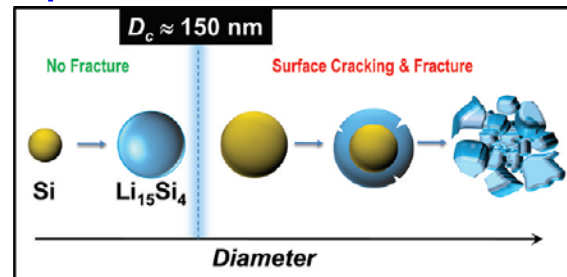
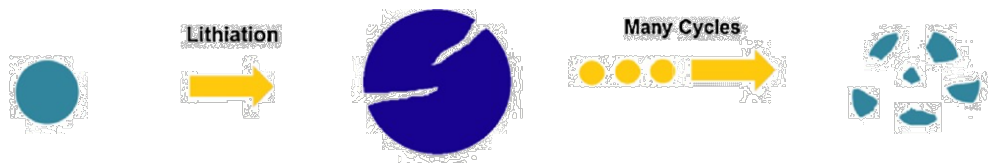
Commercial Graphite (C₆Li)
Gravimetric capacity **371 mAh/g**
Volumetric capacity **810 mAh/cm³**



Issues with Volume Expansion

Smaller particle size

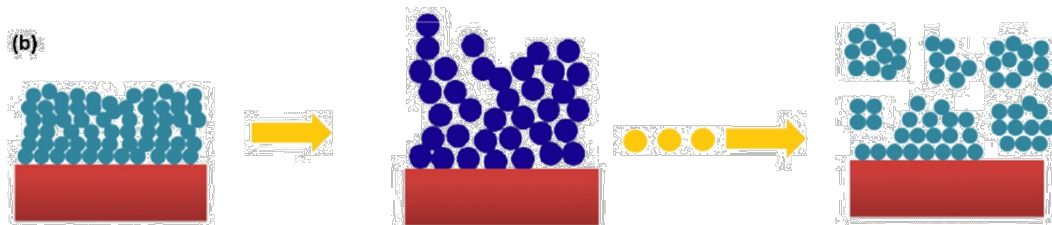
Pulverisation



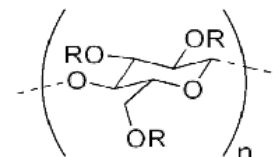
Liu et al, *ACS Nano* **2012**, *6*, 1522.

Flexible binders

Electrode expansion/ impingement

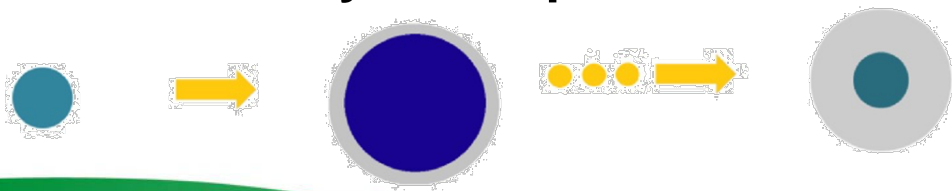


Carboxymethyl cellulose

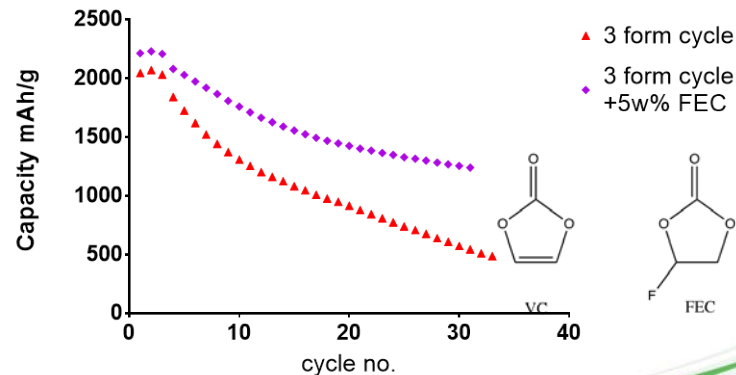


R = H or $\text{CH}_2\text{CO}_2\text{H}$

Solid Electrolyte Interphase formation

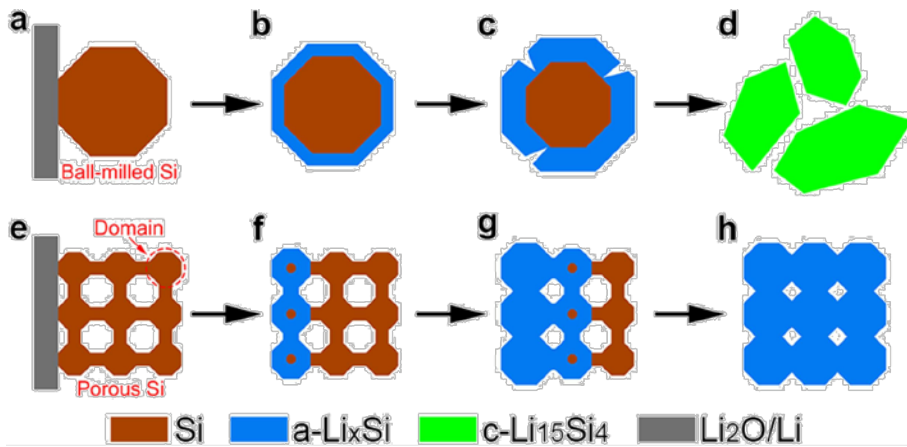


Electrolyte additives

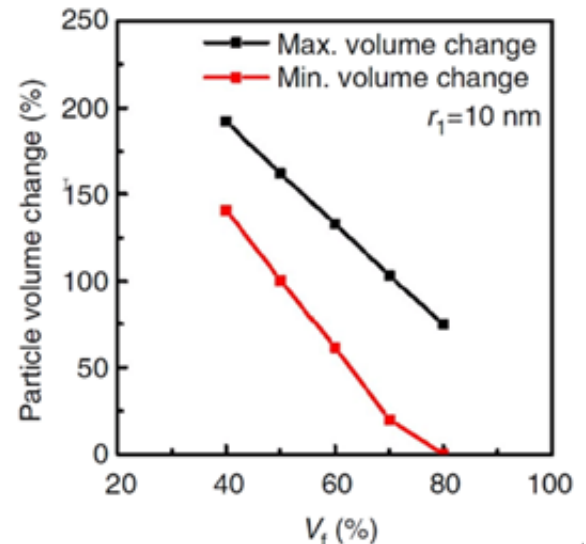


Porous silicon as a solution

- High surface area \Rightarrow better **accessibility** for electrolyte
- Shorter lithium **diffusion** lengths due to small substructures
- Expansion into own pore volume \Rightarrow minimum **cracking**
- More favourable **lithiation** behaviour



Shen et al, *Sci. Rep.*, 2016, **6**, 31334



Li. et al, *Nat. Commun.*, 2014, **5**,
1

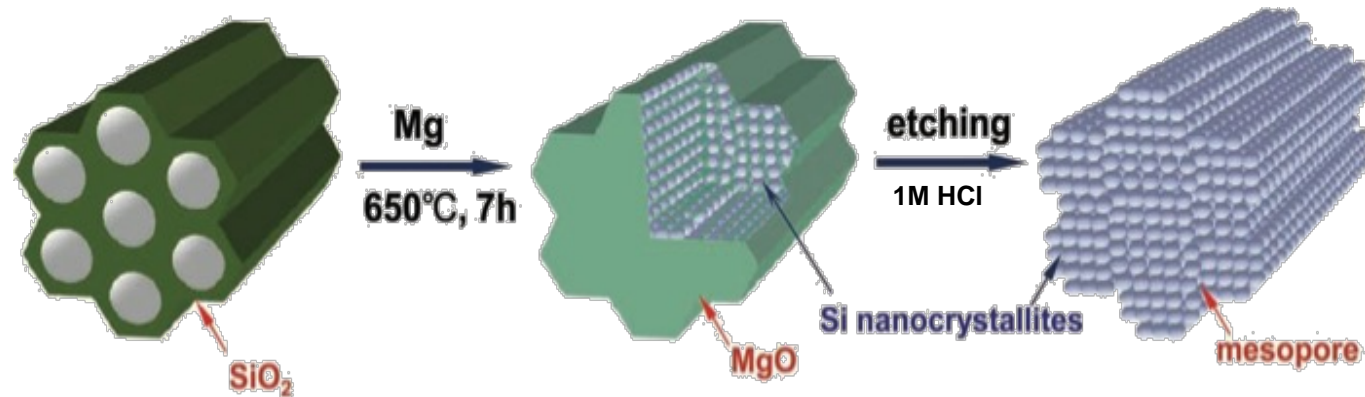
Current Routes to Porous Silicon

- Electroless etching
- Chemical Vapour Deposition
- Electrochemical etching

Most effective for surfaces

For Battery applications, bulk material synthesis routes are needed

Magnesiothermic Reduction for bulk porous silicon production



Good capacities and cyclability:

- 2500 mAh/g for 100s of cycles
 - with <20% capacity loss.



Jia et al. *Adv. Energy Mater.*, 2011, **1**, 1036
Entwistle et al. *J. Mater. Chem. A*, 2018, **6**, 18344

Magnesiothermic Reduction

Missing link: how the reduction process affects the properties and performance of porous silicon?

Solution: A systematic study linking processing-structure-property.



Feedstock

- Silica sources
- Scalability, availability & cost
- Properties of feedstock
-

Process

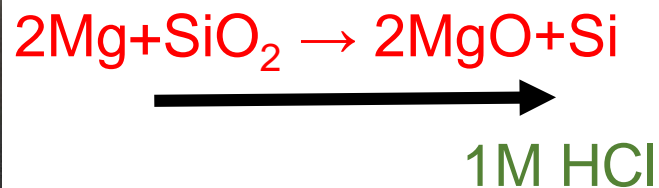
- T and t
- $C_{silica} : C_{Mg}$
- dT/dt
- Thermal moderators (NaCl)
-

Product

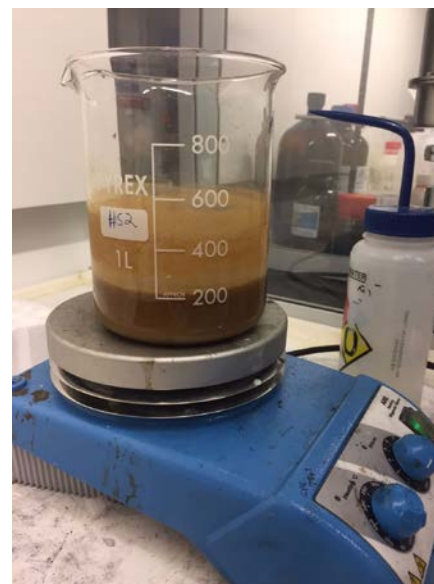
- Yield
- Purity
- Pore structure
- Crystallinity
-



Outline of Magnesiothermic Reaction

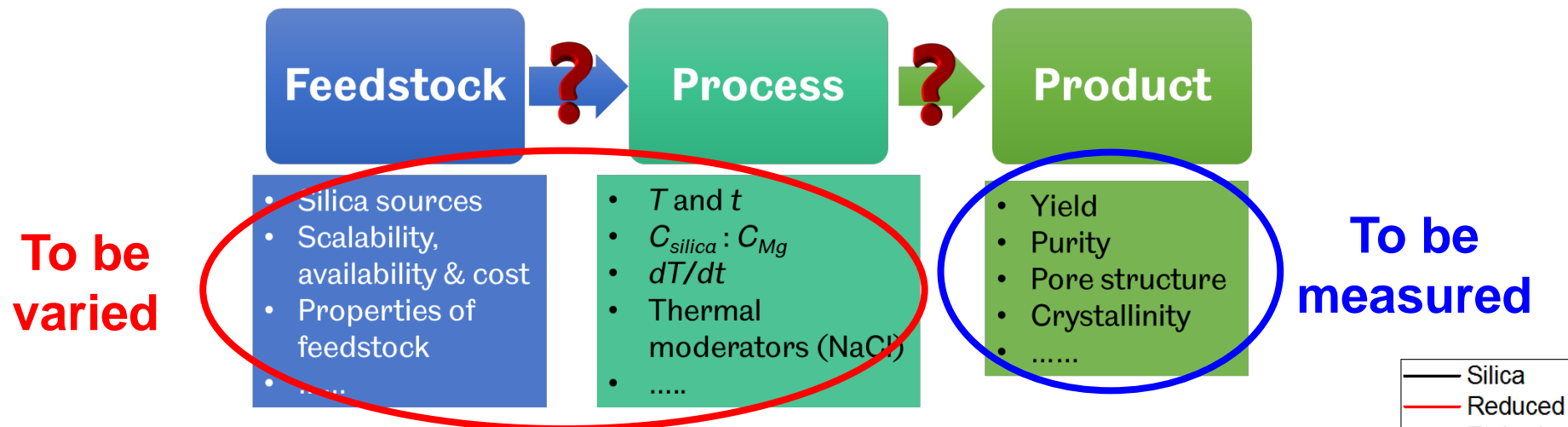


Reduction in a furnace under flowing Ar

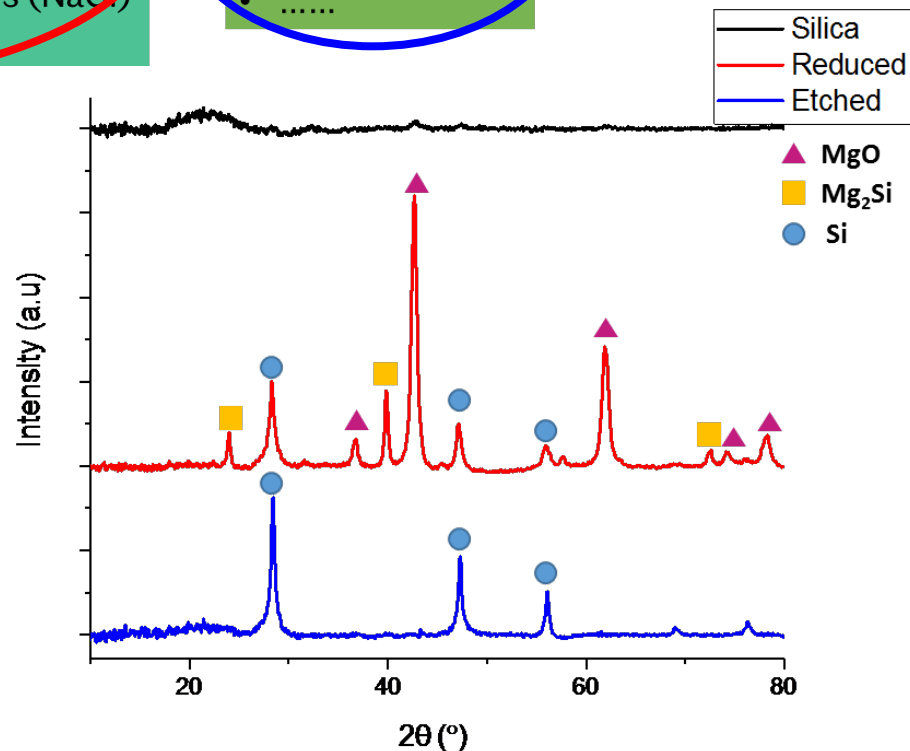


Removal of MgO from porous silicon

Studying the process



- X-ray diffraction (XRD)
- Thermal Gravimetric Analysis (TGA)
- N_2 Porosimetry (surface area and pore sizes)

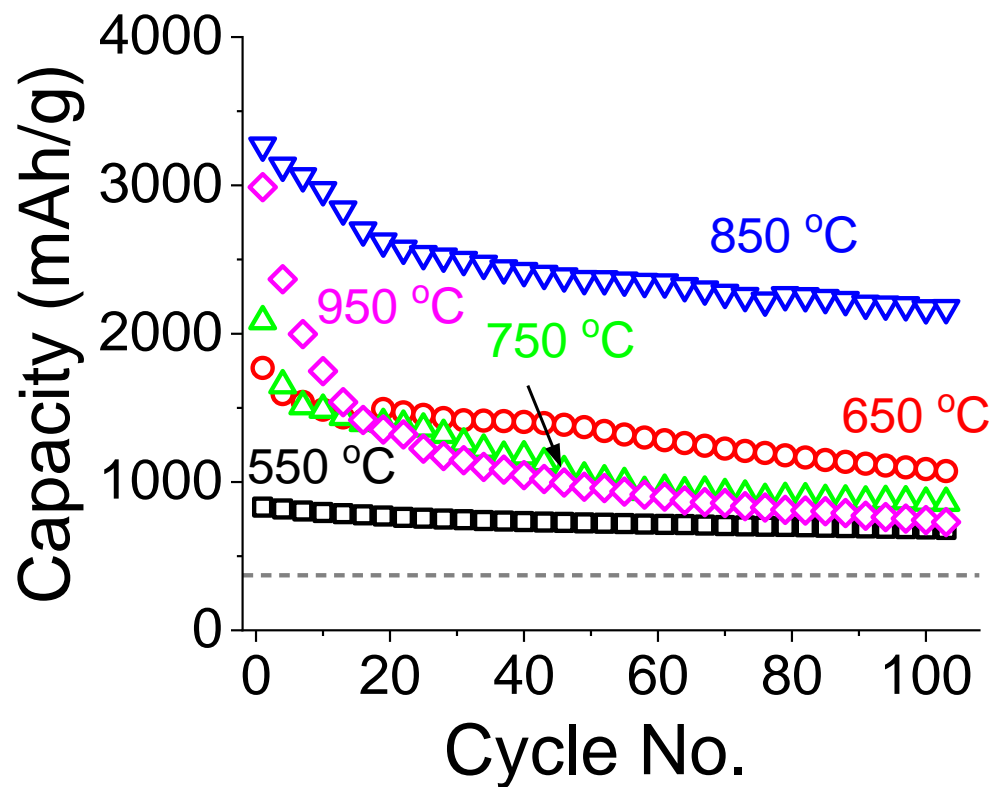


*Publication *Currently Under Review.*

09_2019

Anode Performance

- The capacity retention is very good for some of these samples.
- $\uparrow T$ leads to \uparrow capacity (due to high purity).

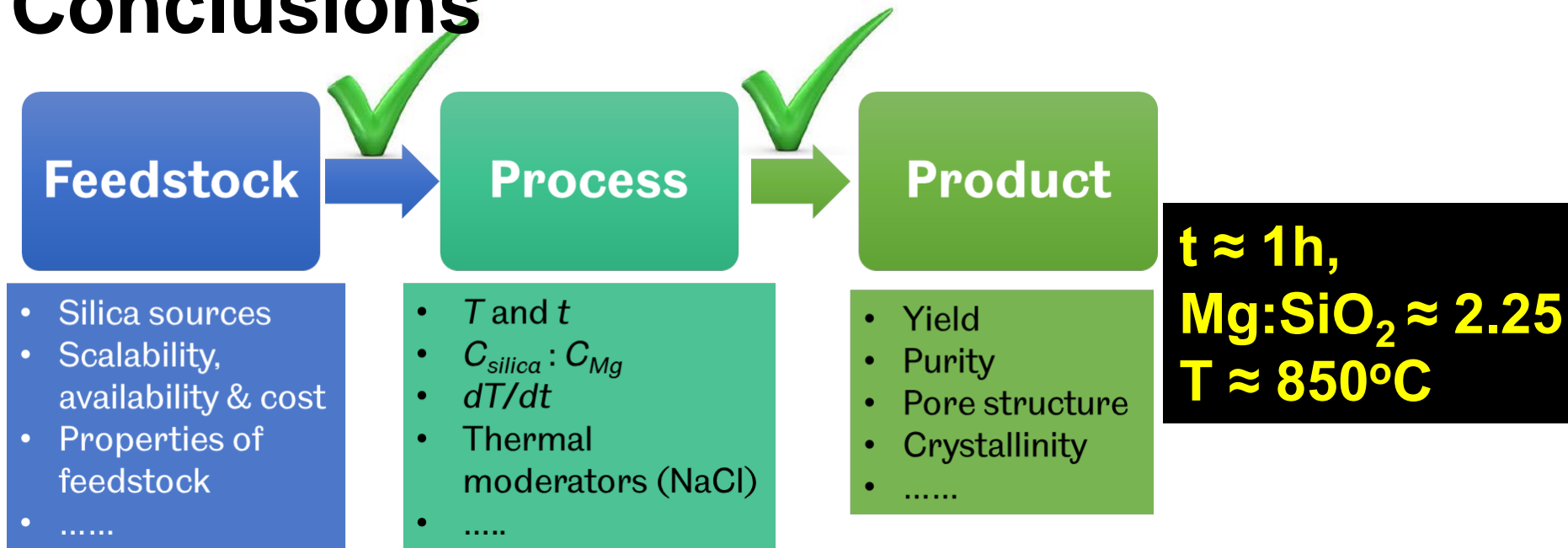


- Porosity is key for extended cycle life span.

**$t \approx 1h,$
 $Mg:SiO_2 \approx 2.25$
 $T \approx 850^\circ C$**



Conclusions



*Publication *Currently Under Review.*
09_2019

Future work

- Optimise the porous silicon for high capacity and stable cycle life
- Techno-economic feasibility of magnesiothermic reduction
- Partnering with industry for large scale manufacturing.



Acknowledgments



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Engineering and Physical Sciences
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Thank you

Application of magnesiothermic reduction of silica to produce porous silicon for lithium-ion batteries

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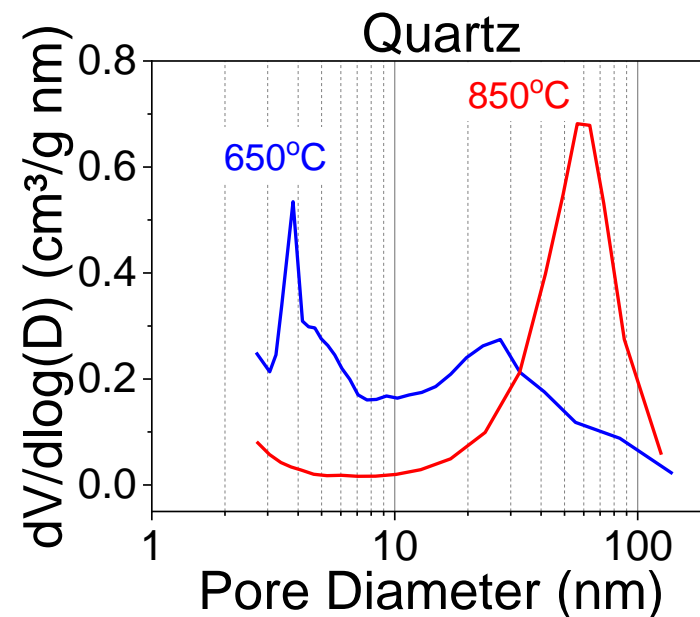
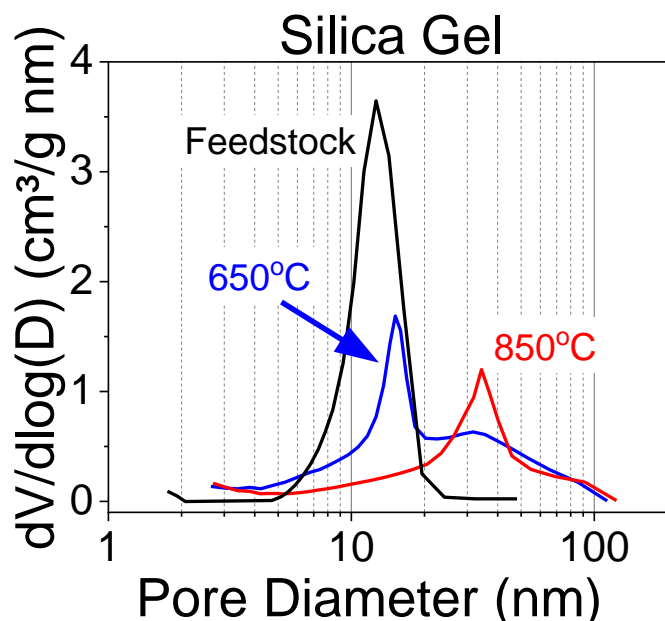
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4. Feedstock....

2 Feedstocks: mesoporous silica and non-porous quartz

2 Ts for Reduction: 650 and 850 °C



- Analysis of pore evolution is new.
- Porosity was introduced into non porous quartz.

....4. Feedstock

